

Formulation and *in-vitro* drug Released Mechanism of CNS Acting Venlafaxine Nanostructured Lipid Carrier for Major Depressive Disorder

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ABSTRACT

Venlafaxine (VLX) is a first line, dual acting and unique antidepressant drug which belong to the class of serotonin and norepinephrine reuptake inhibitors. Oral administration of VLX has many adverse effects, poor bioavailability due first-pass hepatic metabolism and low permeability shows poor antidepressant action in the brain. The aim of this present study was to formulate Venlafaxine Nanostructured lipid carrier (VLX-NLC) and deliver directly into the brain through intranasal route. VLX-NLC were prepared by Melt-Emulsification-Ultrasonication process and characterized by particle size, polydispersity index, zeta potential, Encapsulation efficiency and analyzed by differential scanning calorimetry (DSC), X-ray diffraction (XRD), transmission electron microscope (TEM). The VLX-NLC was also evaluated for *in-vitro* drug release and *ex-vivo* sheep nasal mucosa permeation. The mean particle size of VLX-NLC was between 155 and 293 nm in diameter with entrapment efficacy was up to 74.13%. TEM gave confirm of spherical nature of NLC. The DSC result shows a sharp peak at 208°C corresponds to melting peak hence confirm the peak of Venlafaxine and X-ray diffraction 2θ scattered angle represent crystallinity nature of pure Venlafaxine whereas VLX-NLC reveals decrease intensity of peak which indicates amorphization of VLX due to solubilization in lipid. *In vitro* studies exhibit initial burst release and afterward prolong release, *ex vivo* permeation through nasal sheep mucosa showed 66.45 % of drug diffused in 24 h from VLX-NLC.

Key words: Venlafaxine, Nanostructured lipid carrier, Antidepressant.

INTRODUCTION

Depression is a major depressive disorder which common, serious mental illness and prevents a person from functioning usually. It is related to the high mortality rate.¹ However pathology of mental illness is not known but have a deficiency of norepinephrine and serotonin called as monoamines in the brain may cause depression.² Symptoms including irritability, fatigue, Worry and restlessness, experiences of nervousness, impaired Concentration and sleep disturbance.³ The brain is a delicate organ. Delivery of drugs into the brain is a challenge in the management of neurological disorders due to the presence of blood-brain barrier (BBB).⁴ Blood-Brain

Barrier is a specialized system of capillary endothelial cells that protects the brain from Entry of harmful neurotoxic substances in the bloodstream while Supplying to the brain with the required nutrients for proper function and obstacle to the delivery of active drug constituent into the Central Nervous System (CNS) for disease treatment.^{5,6,7}

Intranasal delivery of drugs results in high blood levels as compared to intravenous route.⁸ Intranasal (IN) drug delivery is needle-free, non-invasive, painless, self and alternative administration route to targeted drugs directly to the brain via

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the olfactory region to achieve faster and higher level of drug absorption by passing the blood-brain barrier shows a rapid onset of action, potential for nose to CNS delivery and no first-pass metabolism.^{9,10,11} Venlafaxine HCL (VLX) is (1-[2-(dimethylamino)-1-(4-methoxyphenyl)-ethyl] cyclohexanol hydrochloride) a first line, bicyclic phenylethylamine derivative of dual action unique antidepressant drug used to treat depression. Its molecular weight is 313.87 with a chemical formula of (C₁₇H₂₇NO₂HCl). Structurally it differs from other Antidepressant which belongs to the class of serotonin and norepinephrine reuptake inhibitors (SNRI).¹² VLX is commercially available in the market as capsules immediately released tablet and controlled release tablets form. VLX is commonly used as oral antidepressants; nevertheless oral administration of VLX has a so many adverse effects such as deliberate onset of action, side effects like a headache, tachycardia, increased blood pressure, dry mouth, dizziness, fatigue, sexual dysfunction and low bioavailability.¹³ Approximately 92 % of VLX absorbed by the oral administration but it undergoes extensive first-pass metabolism in the liver to des-methyl venlafaxine active metabolite. Only 12.6 % is remaining in systemic circulation, hence shows the low onset of action. The solubility of VLX is more than 500 mg in one ml of water and log P value 0.43 which representing hydrophilic nature of drug thereby limiting its permeability and shows poor antidepressant effect in the brain^{14,15,16}. The elimination half-life of Venlafaxine is 4-5 h, therefore, needs to frequent administration of VLX to maintain an effective therapeutic concentration into the brain. The efficacy depends on the continuous availability of the antidepressant drug in a brain. Therefore, sustained release is required to maintain the steady-state levels which are narrow in a case of conventional as well as parenteral remedies.¹⁷ Nanostructured Lipid Carrier (NLC) is the new second generation of solid lipid nanoparticles which consist of the mixture of various solid and oil lipid resulting matrix lipid particles. NLC offers an increased amount of payload of drugs and minimized drug expulsion during storage due to the use of oil lipid over solid lipid nanoparticle. They are biodegradable, biocompatible with Generally Recognized as Safe (GRAS) status, solid at room and body temperature.^{18,19}

The aim of this present study was to deliver venlafaxine HCL into the brain by administering it through the intranasal route in lipid Nano particulate form. VLX-NLCs were developed in a mixture of solid and oil lipids, drug concentration, surfactant and co-surfactant ratio with stirring speed and ultra-sonication time. With the aid of solid and oil lipids were compared with particle

size distribution, polydispersity index, zeta potential and drug encapsulation efficiency. The finished products were analyzed by dynamic light scattering (DLS), differential scanning calorimetry (DSC), X-ray diffraction (XRD), transmission electron microscope (TEM), *In vitro* release studies and Ex-vivo drug diffusion.

MATERIALS AND METHODS

Venlafaxine was obtained as a gift from Ranbaxy research lab Gurgaon (India), Solid-lipid: Geleol pellet (Glyceryl palmitostearate), Gelucire 50/13 (stearoyl macrogol-32 glyceride), Compritol 888 (Glyceryl behenate), Glyceryl Monostearate (monosteric acid ester of glycerol) and Precirol ATO 5 (Glyceryl stearate) were obtained by Gattefosse France as a free gift sample. Oil Lipid: Medium chain triglyceride (MCT) such as Capryol 90 (propylene glycol monocaprylate), Miglyol 808, Captex 200 P, Maisine (glycerol monolinoleae), Capmul MCM, Capryol PGM (propylene glycol II monocaprylate) obtained from ABITEC, Mumbai as a free gift sample. Pluronic F127 was from Sigma-Aldrich. Sodium taurocholate selected as bile salt from S D fine Mumbai. Potassium dihydrogen phosphate, sodium hydroxide (NaOH) was purchased from Merck Mumbai. Dialysis bag MW 12,000 Da Himedia Mumbai India. Millipore water freshly purified (Milli-Q Gradient A10 system, Elix 0.22μ m), freshly prepared double distilled methanol was whenever required. All the reagents and chemicals used of analytical grade.

Screening of lipids for NLC formulation

Solid lipid

Screening of solid Lipid was determined by adding an excess amount of Venlafaxine in 2 g of melted various solid lipids in controlled temperature Remi water bath shaker, Mumbai at 75 °C mixed thoroughly with a vortex mixer (Remi CM 101). The total amount of Venlafaxine solubilized in solid lipids was determined visually by initial saturation.

Oil lipid

Screening of oil Lipid was determined by shaken flask method. The excess amount of venlafaxine was added in 2 g of various oil lipids in screw cap glass bottle. All the bottles were covered with an aluminium foil to avoid entry of vapours in the bottles and placed in thermostat wrist hand water bath shake (Remi Model RSB12), Mumbai for 24 h at 40°C. The supernatant liquid was filtered. After filtration diluted a drop of supernatant in double distilled methanol solubility was analyzed by

UV-Visible spectrophotometer (Shimadzu UV/1800) at λ 276 nm.^{20,21}

Melt-Emulsification-Ultrasonication method followed by lyophilization

The melt-emulsification-ultrasonication process was used for the preparation of Venlafaxine nanostructured lipid carrier (VLX-NLC). In NLC preparation lipid phase and aqueous phases were separately prepared. Accurately weight amount of Glyceryl monostearate as solid lipid and Capmul MCM as oil lipid in a ratio of 7:3 w/w were heated at 5 to 10°C above the melting point of the solid lipid that is 75°C to prevent recrystallization of lipid during the process. The drug was added in the melted lipid to dissolve. Meanwhile, separately aqueous surfactant phase was prepared in Millipore water (Milli-Q Gradient A10 system Elix 0.22 μ m) in that stearic surfactant Pluronic F127 and sodium taurocholate (bile salt) maintain temperature 75°C with continuous stirring on electronic magnetic stirrer (2 MLH Remi Magnetic stirrer), Mumbai. The aqueous surfactant solution was added slowly drop by drop with the help of borosilicate glass syringe to lipid phase under magnetic stirrer at 5 to 10°C above the melting point of the solid lipid. The obtained coarse emulsion was further treated with a probe-type sonicator (Probe ultrasonicated Bandelin sonoplus) in a water bath at 75°C for 15 min at 40 amplitude. Finally, prepared nanostructured lipid carrier was cooled down at room temperatures on a magnetic stirrer (4 EMS Electromagnetic stirrers) in which lipids get re-crystallized to form Venlafaxine loaded nanostructured lipid carrier.^{22,23}

Freeze-drying of NLC dispersion

VLX-NLCs dispersions were lyophilized to obtain formulation in dry form. The obtained VLX-NLC was frozen using freezer at -20°C for overnight, VLX-NLCs were transferred to the lyophilizer at the temperature at -70°C for 48 h. After lyophilization VLX-NLC lyophilized form was collected from lyophilizer and was subjected to physicochemical characteristics, in-vitro, ex-vivo and *in vivo* studies.²⁴

Particle size (PS) and polydispersity index (PDI) measurement

The average particle size (PS) and polydispersity index (PDI) of VLX-NLCs were determined by Photon correlation spectroscopy (PCS) at the detection angle of 90°C at 25°C. The zeta potential was analyzed by a Nano-ZS Zeta Sizer (Malvern Instruments, Malvern, UK Nano

size) at 25°C after proper dilution with double distilled water. Each measurement was made in duplicate.^{25,26}

Drug Excipient Compatibility Studies

Compatibility studies of drug and solid or liquid Lipids material were evaluated to check whether any chemical interaction between them during the formulation of NLC was recorded by Fourier Transform Infrared Spectroscopy FTIR Spectrometer (Shimadzu FTIR-8400 S CE) using KBr pellets (400-4000) with a scanning speed of 2 min/sec with normal slit.

Differential scanning calorimetry (DSC) analysis

The crystalline state of drug samples was recorded with differential scanning calorimetry (DSC). For DSC measurement, accurately weight amount 5 mg of drug samples into aluminium pans and then hermetically sealed with aluminium lids. The thermograms of samples were obtained at a scanning rate of 10°C/min at a different temperature range between 0 and 400°C in a nitrogen atmosphere. The thermal measurement of physical mixture of VLX-Lipid was also carried by differential scanning calorimetry.²⁷

Transmission electron microscope (TEM)

The morphological observation of optimized VLX-NLC was carried out by Transmission electron microscope operated at voltages ranging from 100 to 200 kV. A placing a drop of the sample which was diluted with double distilled water and spread on a 200-mesh copper grid coated with carbon membrane and negatively stained with 2% phosphotungstic acid for 30 s. The grid was dried at room temperature and then observed by TEM.²⁸

X-ray diffraction (XRD)

A crystalline structure of the VLX-NLC was investigated using an X-ray diffractometer (Rigaker Geiger flex, Japan). Aqueous NLC dispersions were lyophilized before to the XRD measurement. Diffractograms were performed from the initial angle $2\theta = 10^\circ$ to the final angle 40° with a Cu K α radiation source. Samples were placed in the glass sample holders and scanned from 2°C to 80°C with a scan angular speed 2h/min at 35 kV operating voltage and 40 mA current. The diffraction spectra were recorded.²¹

Entrapment efficiency of VLX

The amount of drug incorporated in NLC was determined after separation of the free drug and lipids from the formulation. The 3ml volume of venlafaxine nanostructured lipid carrier suspension was transferred to the small centrifuge tubes were kept in ultracentrifuge

(Beckman Coulter Germany). The nanostructured lipid carrier was centrifuged at 70,000 rpm for 20 min. The 1 ml of supernatant liquid was collected and diluted with an appropriate volume of water. The amount of drug was determined by UV-Visible spectrophotometer (Shimadzu UV/1800) at λ 276 nm. The % encapsulation Efficiency (EE) was calculated by the following equations, respectively. All measurement was performed in triplicate.²⁹

$$\% \text{ EE} = \frac{W_{\text{TOTAL}} - W_{\text{FREE}}}{W_{\text{TOTAL}}} \times 100$$

In vitro drug release study

In vitro drug release from Venlafaxine and all the batches of VLX-NLC were performed in 250 ml phosphate buffer solution (PBS) pH 6.4 using paddle Method and vessel assembly from USP type II eight station Dissolution apparatus (Electrolab-TDL-08L). The aqueous VLX-NLC (equivalent to 5 mg Venlafaxine) was placed in dialysis membrane bag (cellophane tube Himedia MW cut off 12,000 Da) which was previously clean, washed and removes impurities. The membrane bag was soaked overnight in PBS pH 6.4. The two ends of dialysis bag were sealed with thread to avoid leakage. The membrane was immersed in a vessel containing 250 ml PBS pH 6.4 thermostatically controlled water bath maintained at $37 \pm 1^\circ\text{C}$, rpm of the paddle was set at 50. At regular time intervals (0.5, 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 12 and 24h) sample was withdrawn and immediately equal volume of fresh PBS was replaced. The concentration of drug was analyzed using UV-Visible spectrophotometer (Shimadzu UV/1800) at a wavelength of λ 274 nm. The release studies were performed in triplicate.²⁹

Determination of drug Released Kinetic

The data obtained after In-Vitro released of selected NLC formulation was determined by zero order, first order, Higuchi matrix, and Peppas Korsmeyer models.^{30,31}

Ex vivo drug release study

Ex vivo drug diffusion study was performed on nasal tissue specimen of sheep for predicting the ex-vivo drug release characteristics. Freshly excised sheep nasal tissue specimen was collected from local slaughter, immediately dipped in freshly prepared phosphate buffer solution (pH 6.4). The bony cartilage was removed from the mucosal membrane, nasal tissue was isolated, washed and stabilized under phosphate buffer solution (pH 6.4). The tissue was stabilized in donor and acceptor compartments under phosphate buffer solution (pH 6.4). In *ex vivo* diffusion study 1ml Venlafaxine pure drug solution was introduced (2 mg ml⁻¹) on freshly sta-

bilized nasal mucosa tissue which placed in the donor compartment. The receptor compartment was filled with 50 ml phosphate buffer solution (pH 6.4). The stirring speed was 200 rpm with a magnetic stirrer at $37 \pm 5^\circ\text{C}$. In the case of Venlafaxine nanostructured lipid carrier and the lyophilized VLX-NLC equivalent amount to 2 mg of Venlafaxine was reconstituted with 1 ml phosphate buffer solution (pH 6.4). Aliquot 2ml of Samples at specific time intervals (0.5, 1, 2, 4, 6, 8, 10, 12 and 24 h) and immediately replaced an equal volume of fresh buffer solution (pH 6.4). The amount of drug diffusion was analyzed using a UV-Visible spectrophotometer (Wensler) at wavelength λ 274 nm and % drug diffusion was calculated based on the standard calibration curve of Venlafaxine.^{29,32,33}

Stability Investigation

As per the stability of NLC, concerned lyophilized VLX-NLCs was placed in screw cap glass bottle which covers with aluminium foil stored in refrigerator temperature approximately $4-8^\circ\text{C}$ and room temperature at 25°C under for 1-3 month. After 1-3 month NLCs lyophilized formulations were analyzed for particle size and EE and then compare with new formulations.^{34,35}

Statistical analysis

All experiments were performed in triplicate and data were reported as mean \pm SD (standard deviations) in Microsoft office excel.

Results and discussion

Lipid Screening

The various solid lipid material eg, Geleol pellet (Glyceryl palmitostearate), Gelucire 50/13 (stearoyl macrogol-32 glyceride), Compritol 888 (Glyceryl behenate), Glyceryl Monostearate (monosteric acid ester of glycerol) and Precirol ATO 5 (Glyceryl distearate) and Medium chain triglyceride (MCT) such as Capryol 90 (propylene glycol monocaprylate), Miglyol 808, Captex 200 P, Maisine (glycerol monolinoleae), Capmul MCM, Capryol PGMC (propylene glycol II monocaprylate) are used as for screening of lipid in NLC production. The selection of lipid was done carefully as per the solubility of drug in lipids. Figure 1a and 1b showed that a higher amount of VLX could be solubilized in Glyceryl monostearate and Capmul MCM gave 179.5 mg/g and 516 mg/g. Therefore solid lipid Glyceryl monostearate and oil lipid Capmul MCM was selected on the base of higher solubility of drug for the NLC development. By visual examination of blending a Glyceryl monostearate and Capmul MCM in ratio of 70:30 was found that no any oil droplets were seen on glass slide under microscope.

The solubility data of various solid and oil lipid was shown in Table 1. The selection of surfactant such as Pluronic F127 and sodium taurocholate (bile salt) was selected on the basis of Hydrophilic-lipophilic balance (HLB) values and possess physicochemical properties which suitable for nose to brain drug delivery system, Inhibit P-glycoprotein and multidrug resistance-associated protein efflux, enhancing the permeability of drug across the nasal mucosa and BBB.^{12,34}

Formulation and characterization of Venlafaxine nanostructured lipid carrier

VLZ-NLCs were prepared by melt emulsification ultrasonication followed by lyophilization. In this present study, Glyceryl monostearate was used as solid lipid and Capmul MCM was used as oil lipid. Pluronic F127 was used as Stabilizer and sodium taurocholate (Bile salt) as co-surfactant. All NLC formulation was prepared at 75 °C to stay the solid lipid remains in the liquid state throughout the production process and do not recrystallized during the process. The composition of the formulation as shown in Table 2. The particle size distributions of VLX-NLC was found to be 199.6 nm shown in Figure 2. The mean particle size of the all VLX-NLC dispersion was varied between 155 and 293 nm with a polydispersity index value of all was below 0.7, representing narrow and uniform particle size distributions depending on the concentration of Pluronic F127, Sodium taurocholate and stirring speed. VLX-NLC

4 was chosen as best VLX-NLC formulation because of optimum mean particle size, PDI and zeta potential of were 213 ± 4.3 , 0.49 ± 0.015 , -35 mV and respectively. The selected VLX-NLC with zeta potential was as shown in Figure 3. A favorable concentration of Pluronic F127 which leads to decreasing the particle size of NLC. Hence, the concentration of Pluronic F127 (1% w/v) was selected. However, a decrease in particle size with increased concentration of surfactant was used. Polydispersity index is determining particle homogeneity varies from 0 to 1. Closer the value of zero polydispersity is better the homology between the particles sample. The particle size of Venlafaxine NLC formulation was further confirmed by TEM. The surface charge of the particles was measured by Zeta potential (ZP) which indicates the degree of repulsion and predicts the physical stability of the NLC dispersions. Zeta potential of Venlafaxine dispersion was -34 . The decreases in zeta potential due to the presence of a free drug in the NLC dispersion which is anionic in nature whereas used of cationic surfactant in formulation ensuring in enhanced positive zeta potential. Owing to Hydrophobic nature of the solid lipids and oil. It is difficult to encapsulate and preserve hydrophilic drugs in the lipid matrix. However, to overcome this problem, variety of the formulation excipients was found to be crucial. Arpana Patil Gadhe and Varsha Pokharkar have improved the loading of hydrophilic drug Montelukast in Nanostructured Lipid Carrier.²⁰ the Encapsulation Efficiency of

Table 1: Solubility studies of Venlafaxine in various solid and Oil lipids. (n=2)

Solid Lipids	solubility (mg)	Oil Lipids	solubility (mg)
Glyceryl Monostearate	179.5	Captex 355	152
Stearic Acid	154	Capryol 90	396
Compritol 888	70	Capmul MCM	516
Precirol ATO 5	74.5	Capmul PG 8	509.5
Gelucire 50/13	75.5	Captex 200 P,	60.9

Table 2: Formulation Composition of VLX- NLC

Batches	GMS (mg)	Capmul MCM	Drug (mg)	F127 (%) (mg)	STC (%) (mg)	Stirring Speed rpm/35min
VLX-NLC 1	280	120	10	250	125	1000
VLX-NLC 2	280	120	10	500	250	1000
VLX-NLC 3	280	120	10	250	250	2000
VLX-NLC 4	280	120	10	500	125	2000
VLX-NLC 5	280	120	10	250	250	1000
VLX-NLC 6	280	120	10	500	125	1000
VLX-NLC 7	280	120	10	500	250	2000
VLX-NLC 8	280	120	10	250	125	2000

hydrophilic VLX showed maximum solubility in Glyceryl Monostearate and Capmul MCM was found to be 174 mg/g and 559 mg/g. The loading of a drug was found to be 27%. The addition of solid lipid (Glyceryl Monostearate), Oil lipid (Capmul MCM) MCT increases in encapsulation efficiency. The % EE can be improved by improving the amorphous nature of particle resulting in enhanced stability and continuous release of a drug. The mean particle size (PS), polydispersity index (PDI), Zeta potential (ZP), % Entrapment efficiency (EE) and Drug loading (DL) of VLX-NLC are shown in Table 3. Each value represented the mean \pm SD (n=3).

Fourier transforms infrared spectroscopy (FTIR)

Figure 4 shows FTIR spectrum of VLX, VLX-NLC and Physical mixture of drug-lipid. The characteristic peak at 3350.35 cm^{-1} of O-H, 1614.42 cm^{-1} due to aromatic C=H stretching, 1514.12 cm^{-1} due to C-O stretching, 1178.51 cm^{-1} due to C-N stretching, 1043.49

cm^{-1} due to C-O-C stretching at, 1732.08 cm^{-1} due to C=O stretching of an important functional group. These characteristic stretching bands were a little different after pre-formulation study and illuminating no Chemical interaction between the VLX and lipids material during the formulation of NLC was observed in IR pattern.

Differential scanning calorimetry (DSC) analysis

The DSC thermograms of pure drug exhibited a single sharp endothermic peak at 208.52°C, related to melting point and representing crystalline nature of the pure drug. The physical mixture of Venlafaxine also showed a peak at 193.87 °C on same thermal behaviour. The sharp peak at 208°C corresponds to melting peak of Venlafaxine, which is not present in the thermogram of VLX-NLCs shown in Figure 5. This indicates either Venlafaxine is completely solubilized in the lipid matrix or transformation to amorphous form when compare

Table 3: The mean PS, PDI, Zeta potential, % EE and in-vitro drug released.

Batch no	Particle size (nm)	PDI	Zeta potential (mV)	E.E. %
VLX-NLC 1	196.8 \pm 4.3	0.446 \pm 0.053	-43.6 \pm 0.74	67.94 \pm 4.2
VLX-NLC 2	115.6 \pm 0.35	0.616 \pm 0.006	-34.32 \pm 1.2	64.19 \pm 3.07
VLX-NLC 3	250.8 \pm 4.4	0.439 \pm 0.016	-32.7 \pm 0.9	70.0 \pm 3.61
VLX-NLC 4	213.7 \pm 3.49	0.49 \pm 0.015	-36.6 \pm 2.3	75.97 \pm 2.10
VLX-NLC 5	289.4 \pm 5.2	0.479 \pm 0.009	-31.02 \pm 2.1	73.48 \pm 4.8
VLX-NLC 6	258 \pm 6.9	0.455 \pm 0.057	-49.4 \pm 1.9	69.65 \pm 6.11
VLX-NLC 7	73.41 \pm 2.33	0.633 \pm 0.10	-33.35 \pm 0.08	66.03 \pm 6.94
VLX- NLC8	229.6 \pm 3.71	0.271 \pm 0.016	-42.04 \pm 2.6	74.13 \pm 5.19

Table 4: Kinetic analysis of optimized VLX-NLC dispersion.

Time (h)	Square root of time	Log time	% Drug released	Fraction drug released	Log fractioned drug released	% drug remaining	Log % drug remaining
0	0.00	0.00	0.00	0.000	0.000	100	2.000
1	1.00	0.00	6.7217	0.067	-1.173	93.27	1.969
2	1.41	0.30	18.3838	0.183	-0.737	81.62	1.911
3	1.73	0.47	25.2004	0.252	-0.598	74.8	1.873
4	2	0.60	33.7304	0.337	-0.472	66.27	1.821
5	2.36	0.69	45.6671	0.456	-0.341	54.34	1.735
6	2.44	0.77	56.0271	0.560	-0.251	43.98	1.643
7	2.64	0.84	68.1321	0.681	-0.166	31.87	1.503
8	2.83	0.90	78.6654	0.786	-0.104	21.34	1.329
9	3	0.95	80.9437	0.802	-0.095	19.06	1.280
10	3.16	1.00	83.2403	0.832	-0.079	16.76	1.224
12	3.46	1.07	85.5453	0.855	-0.068	14.46	1.160
24	4.89	1.38	89.5337	0.89.5	-0.048	10.47	1.019

Table 4a: Co-efficient correlation of optimized VLX-NLC Dispersion.	
Released Profile model	Regression Value(R ²)
Zero Order	0.958
First order	0.974
Higuchi Model	0.854
Peppas Model	0.417

Table 5: Stability profile of optimize VLZ-NLC.						
Parameters	4-8°C Month			25°C Month		
	1	2	3	1	2	3
EE (%)	71.97	71.43	71.62	71.97	70.12	68.82
PS(nm)	199.1	199.1	199.1	199.1	203.02	205.41

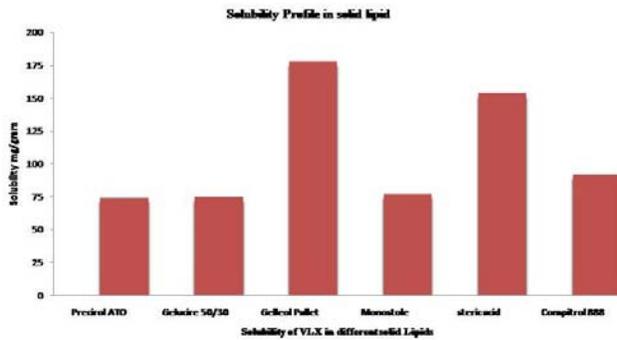


Figure 1 A: Solubility profile of Venlafaxin in solid lipid.

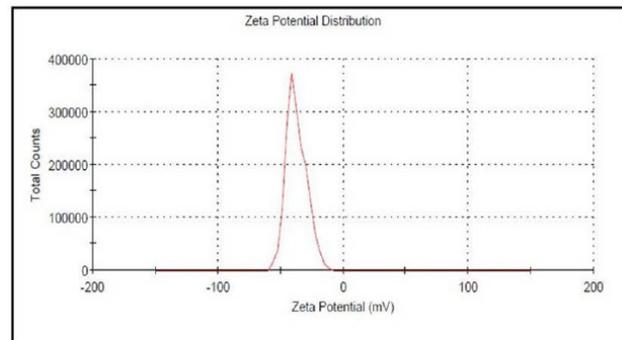


Figure 3: Zeta potential of optimized VLX-NLC .

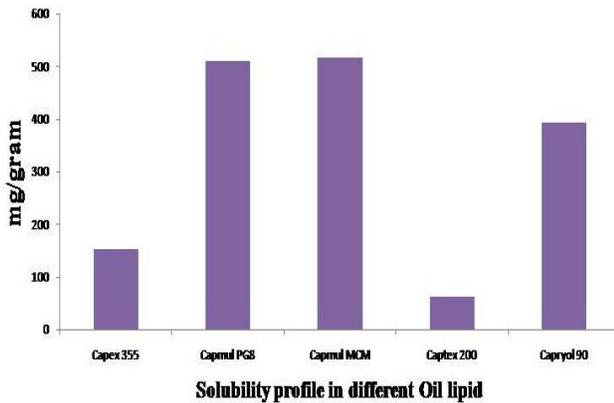


Figure 1 B: Solubility profile in different Oil lipid.

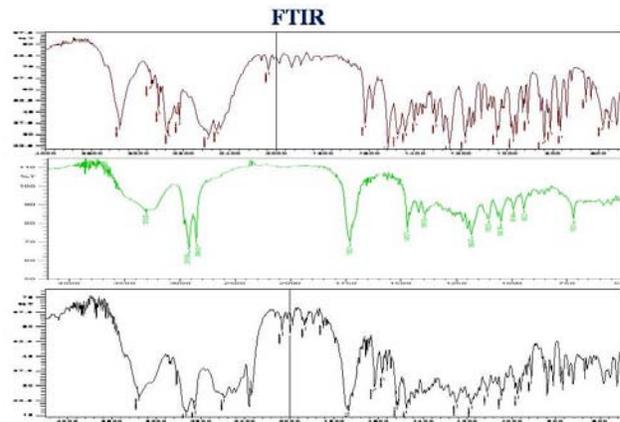


Figure 4: FTIR spectra of venlafaxine, bulk lipid, drug lipid mixture.

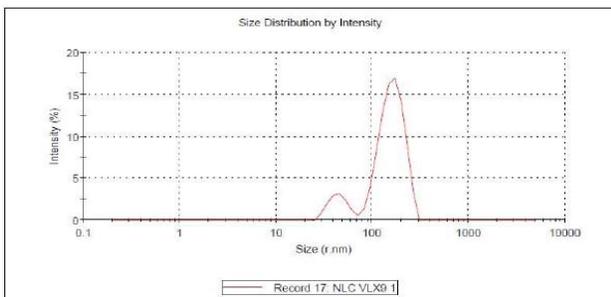


Figure 2: Particle size distribution of optimized VLZ-NLC.

to VLX alone. DSC profile of NLCs was investigated and analysis made up of Glyceryl monostearate and portion amount Capmul MCM in the ratio of 70:30 used in NLC formulation influences the crystallinity of Glyceryl monostearate hence shows broadening of the peak.

X-ray diffraction (XRD) investigation

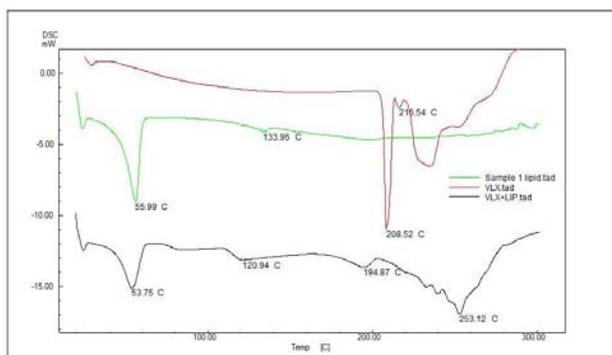


Figure 5: DSC image of 1. VLX 2. Lipid and 3 drug lipid mixture.

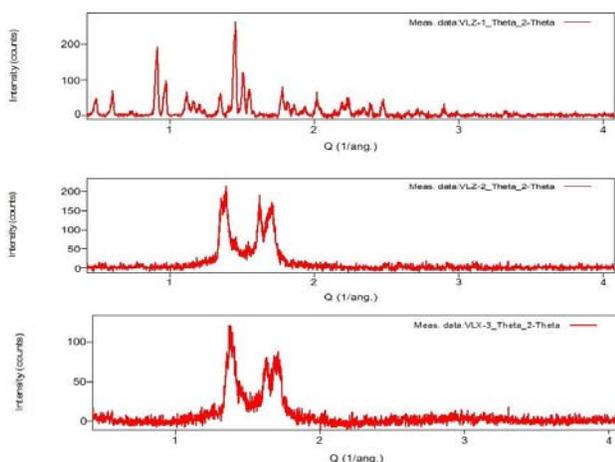


Figure 6: X-ray diffraction analysis, 1) Venlafaxine, 2) VLX-Lipid-PM, 3) VLX-NLC.

The crystallinity study of Venlafaxine (VLX), VLX-PM (physical mixture) and VLX-NLC formulation (lyophilized) was performed by X-ray diffraction. Figure 6 shows XRD of Venlafaxine, VLX-PM (physical mixture) and VLX-NLC formulation (lyophilized). The diffractograms of Venlafaxine show the sharp peak in 2θ scattered angle representing crystallinity nature of Venlafaxine whereas physical mixture and VLX-NLC formulation show decrease intensity of the peak of Venlafaxine which indicates amorphization of Venlafaxine.

Transmission electron microscope (TEM)

TEM images of optimized VLX-NLC formulation which represented the particles size were spherical in nature. The diameter of the VLX-NLC formulation was small 200 nm shown in Figure 7. TEM provides two-dimensional morphological images and capability to detect the size of the sample in the nanometer range. TEM is capable to form the images with higher resolution than SEM. The size difference obtained by TEM and DLS mainly came from the different sample preparation processes and different principles. The DLS did not 'measure' the particle sizes but detected the light scattering

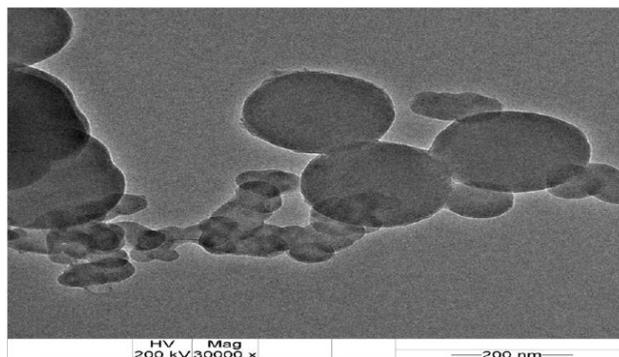


Figure 7: TEM image of optimized VLX-NLC.

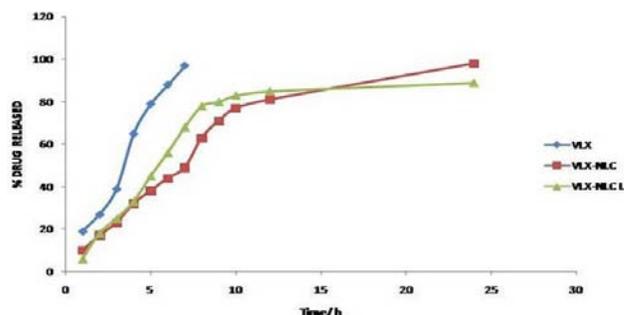


Figure 8: *In-vitro* drug released of optimized VLX-NLC.

effects which were used to calculate the particle size. The diameter from DLS would be a little bigger than that observed from TEM.

In-vitro drug release assay

The *In-vitro* drug release study of Venlafaxine from NLC formulation was studied using dialysis bag method and was compared with Venlafaxine drug solution. *In vitro* drug release profile of VLX-NLC was carried out in phosphate buffer solution (pH 6.4) due to partial solubility in the buffer solution which was published in the previous literature. The % drug was released within 2h from pure venlafaxine solution was found to be 93%, whereas 89.5 percentage cumulative drug was released in 24 h from Venlafaxine nanostructured lipid carrier indicating prolonged released, shown in Figure 8. It was concluded that the released rate from nanostructured lipid carrier could be significantly prolonged the released of Venlafaxine and observed that the rate of release of drug was decreased with a particle size of NLC was increased.

Mechanism of drug released

The data were obtained from *in-vitro* drug release studies were fitted to kinetic model such as zero order, first order, Higuchi square root of time and Korsmeyer-Peppas

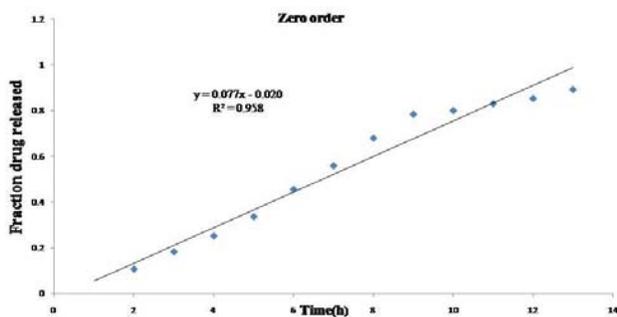


Figure 9: Zero order released form optimized VLX-NLC.

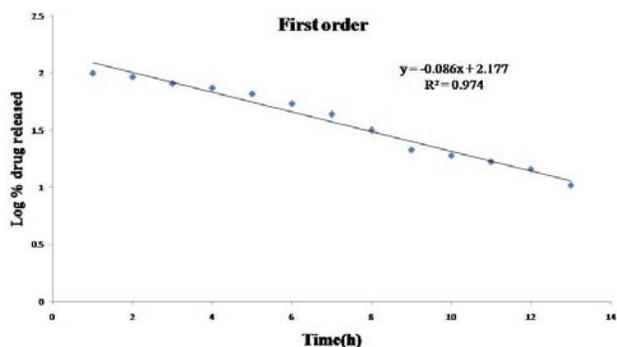


Figure 10: First order released form optimized VLX-NLC.

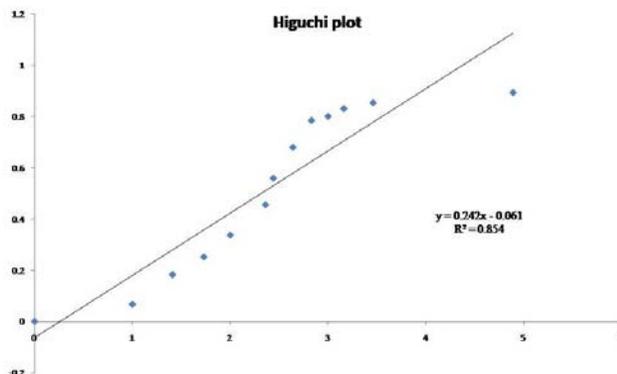


Figure 11: Higuchi model optimized VLX-NLC with size value.

model to know the mechanism of drug release from the VLX-NLC was found to be 0.958, 0.974, 0.8541 and 0.417, respectively were shown in Table 4, and Figure 9,10,11,12 respectively.

Ex- vivo diffusion studies

The ex-vivo drug permeation studies were performed in sheep nasal mucosa for the VLX solution, Lyophilized VLX-NLC and VLX-NLC dispersion. Ex-vivo drug release data of pure drug shows 89 % drug release at 8 h (pH6.4), VLX-NLC exhibit 73.27 % drug release and VLX-NLC suspension plotted graph 76.82 % drug diffused versus time (h) shown in Figure 13 and permeation data was given in Table 6. The pKa value of VLX is 9.4, thus at nasal mucosa pH (5.5-6.5), a large amount of the drug remains in protonated form and

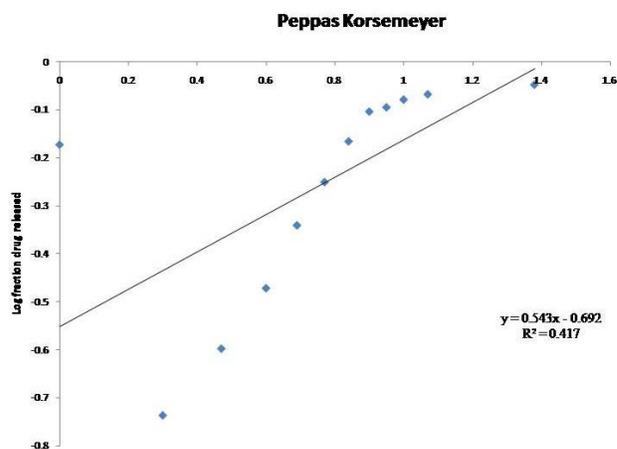


Figure 12: Peppas Model form optimized VLX-NLC.

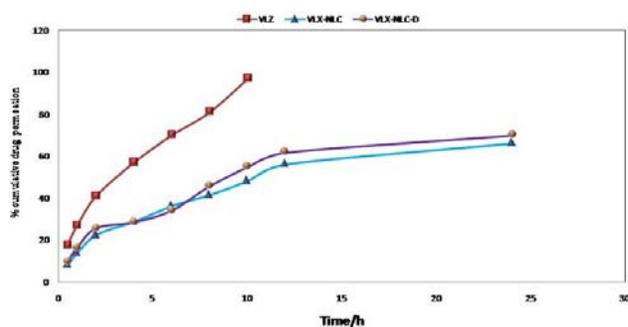


Figure 13: Ex vivo drug diffusion profile from 1) VLX pure drug, 2) VLX-NLC 3) VLX-NLC-Dispersion.

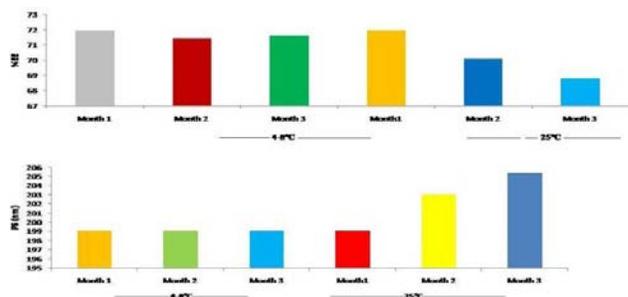


Figure 14: EE and PS of Stability data of optimized VLX-NLC.

so it is transported through tight junctions opened by sodium taurocholate. Therefore, enhance in permeation through nasal mucosa due to an involvement of paracellular transport through tight junctions.^{31, 32}

Stability profile of VLX-NLC

The stability study was performed for finally optimized formulations at 4-8°C and 25°C for the period of three months. The increasing mode of particle size (PS) and polydispersity index (PI), and decreasing trend drug encapsulation efficiency (EE) were observed with storage time at 25°C. Stability study profile indicated that the

formulations (VLX-NLC) have remained stable for a period of more than three months at normal temperature shows in Figure 14. Hence VLX-NLC can be considered as stable formulation.³⁶

SUMMARY AND CONCLUSION

In India Depression is a most common chronic, potentially debilitating psychiatric disorder. Depression is known as major mood disorder. Various types of antidepressants drugs are used considering their effects on neurotransmission and neurotransmitters. The Venlafaxine are taken for study which is selective serotonin reuptake inhibitors (SSRI) drug used for as mood stabilizing properties and used in treatment of psychotic depression. Therefore, the aim and objective of the present study were developed and prepared VLX-NLCs by Melt emulsification and Ultrasonication methods. The obtained VLX-NLC were characterized by Photon correlation spectroscopy (PCS), Transmission electron microscope, particle size, zeta potential, drug loading, Encapsulation Efficiency, X-ray diffraction, *in-vitro* drug release studies and stability study. The *in-vitro* release from VLX-NLCs and mechanism of kinetic. The release rate mostly depends on the drug, solid and liquid lipid interactions. The *ex-vivo* permeation studies showed controlled and prolonged released. The prepared NLCs were checked for their stability in 4-8°C and 25°C conditions for 3 months and the stability showed that the prepared NLCs of venlafaxine hydrochloride are stable formulations.

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COMPETING INTERESTS

The authors declare that they have no competing interests.

CONFLICT OF INTEREST

The authors declare that they have no conflict interests.

ABBREVIATIONS USED

NLC: Nanostructured lipid Carrier, BBB: Blood Brain Barrier, IN: Intranasal, PS: Particle Size, VLX: Venlafaxine,

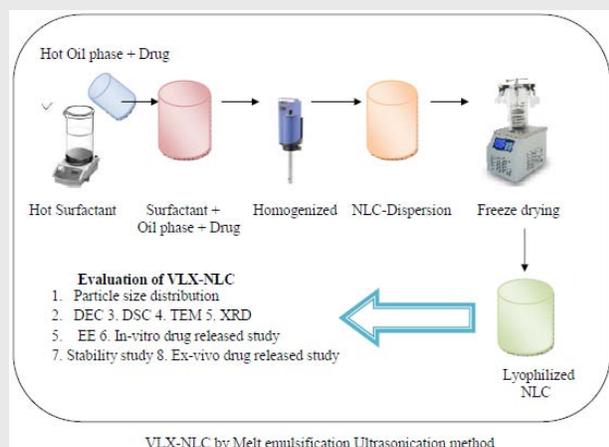
DLS: Dynamic Light Scattering, DSC: Differential scanning calorimetry, XRD: X-ray Diffraction, TEM: Transmission Electron Microscopy, MCT: Medium Chain Triglyceride, PDI: Polydispersity Index, EE: Encapsulation Efficiency,

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PICTORIAL ABSTRACT



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